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JNO₂ at high solar zenith angles in the lower stratosphere

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Abstract. In situ measurements of NO, NO₂, O₃, HO₂, ClO, pressure, and temperature have been made at high solar zenith angles (SZA, 70° - 93°) in the lower stratosphere. These measurements are used to derive the photolysis rate of NO₂, J_{NO₂}, using a time-dependent method. The resultant J_{NO₂} values and the results of a multiple-scattering actinic flux model show a linear relationship throughout the SZA range. The difference of the two sets of J_{NO₂} values of about 11% suggests that the model scattering calculation is very accurate at high SZA conditions near sunrise and sunset.

Introduction

Calculated photolysis rates are an essential part of stratospheric photochemical modeling. Their accuracy depends on both calculated solar actinic flux modeling and laboratorymeasured photolysis cross sections. Validation is often partially achieved through in situ solar flux measurements as a function of SZA. The photolysis rate of NO₂, J_{NO₂}, is of particular interest because it can be derived photochemically from a small set of in situ measurements: NO, NO2, O3, ClO, BrO, and HO2. In previous studies, the agreement between photochemically derived and radiative model JNO2 values for SZAs < 85° was within 8% [Gao et al., 1997] and less than 15% [Del Negro et al., 1999] in the lower stratosphere. The agreement is well within the combined uncertainties associated with the measurements, model calculations, reaction rate coefficients, absorption cross sections, and photolysis quantum yields. These intercomparisons provided an important evaluation of the accuracy of J-value calculations and covered a wide range of parameters (solar zenith angles (SZAs), surface albedo, terrain/cloud height, pressure, and temperature) that affect actinic flux [Swartz et al., 1999]. However, because of this wide coverage, it is difficult to test individual factors. Fur-

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thermore, the intercomparisons were limited to SZAs of less than 85° due to the use of a steady state approximation for the derived J_{NO2} [Del Negro et al., 1999 and references therein].

In this paper we present a time-dependent method for deriving photochemical J_{NO2} values at SZAs beyond 85° in the lower stratosphere (19 - 20 km). The method makes use of in situ measurements of both NO₂ and NO during four stratospheric flights of the NASA ER-2 high-altitude aircraft [Del Negro et al., 1999]. These flights include high SZAs (89° - 93°) in the period before sunset or after sunrise. The analysis is simplified by nearly constant values of latitude, pressure (p), temperature (T), and O₃ for these periods. We show that under these conditions the precision of the derived J_{NO2} values depends only on the precision of the NO/NO₂ ratio which is about 7 to 20% for the results presented here. A comparison of photochemically derived and model J_{NO2} values indicates that the calculation of direct and scattered light at high SZAs is accurate in the 310 to 410 nm wavelength range in the lower stratosphere.

Calculations of JNO2

Photochemical method. The basic photochemical relationship between NO and NO₂ is governed by the chemical reactions:

$$NO_2 + hv \rightarrow NO + O$$
 (R1)

$$NO + O_3 \rightarrow NO_2 + O_2 \tag{R2}$$

$$NO + ClO \rightarrow NO_2 + Cl$$
 (R3)

$$NO + BrO \rightarrow NO_2 + Br$$
 (R4)

$$NO + HO_2 \rightarrow NO_2 + OH$$
 (R5)

Typical relaxation time between NO and NO₂ is about 100 s. Other reactions involving NO_x (= NO + NO₂) are ignored since their photochemical lifetimes are a few hours or longer [Kawa et al., 1992]. Reactions (R1) - (R5) contribute to changes in NO₂ through:

$$d[NO2]/dt = (k2[O3] + k3[ClO] + k4[BrO] + k5[HO2])[NO]$$

$$- JNO2 [NO2]$$
 (1)

where the square brackets denote concentrations, and k_n is the rate coefficient for reaction (Rn) [DeMore et al., 1997]. Equation (1) can be solved for J_{NO_2} as:

$$J_{NO_2} = ((k_2[O_3] + k_3[CIO] + k_4[BrO] + k_5[HO_2])[NO] - d[NO_2]/dt)/[NO_2].$$
 (2)

In the steady state approximation, d[NO₂]/dt is set to zero. This approximation is valid in general only at SZAs less than 85° [Webster, 1987; Kawa et al., 1990]. Between 85° and sunrise or sunset (~94° at 20 km), solar flux changes too rapidly for this approximation to be valid.

The flights used here occurred in April and September 1997 near Fairbanks, Alaska (65°N). In situ flight measurements include NO, NO₂, O₃, HO₂, ClO, p, and T. The treatment and sources of the flight data are the described in *Del Negro et al.*

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Table 1. Mean and Standard Derivation of Measured O₃, Latitude, p, and T Values from Four Flights of the ER-2 Aircraft.^a

Measurement	April	September	September	September
	30, 1997	11, 1997	14, 1997	15, 1997
O ₃ (ppmv) ^b NO _x (ppbv) ^c Latitude (°) p (hPa) T (K)	2.5 (0.14)	1.8 (0.04)	1.8 (0.14)	1.8 (0.08)
	0.59 (0.05)	0.93 (0.03)	0.64 (0.07)	0.62 (0.05)
	64.0 (0.15)	64.4 (0.07)	64.4 (0.08)	64.6 (0.10)
	65.3 (1.12)	62.2 (0.58)	61.8 (1.25)	60.8 (0.77)
	227 (0.8)	222 (0.4)	219 (0.7)	219 (0.5)

^aData set corresponds to points shown in Figure 3. The first number is the mean and the second is the standard deviation.

[1999]. Table 1 gives the mean and standard deviation of O₃, NO_x, latitude, p, and T values from each flight. Data are not available for HO₂ on September 14 and 15, or for BrO during any of the flights. Although the contributions of the BrO and HO₂ terms in (2) are less than 3% and 1%, respectively, HO₂ and BrO values from a photochemical steady state model [Salawitch et al., 1994; Wennberg et al., 1999] are used for completeness.

In the time-dependent analysis presented here, d[NO₂]/dt is derived from the NO and NO2 time series made in the SZA range of 88° to 93° near both sunrise and sunset. In the time series, however, every 10-s data point effectively represents a different air parcel due to the aircraft's high speed (~ 200 m/s) and atmospheric variability. To use NO2 measured from such a time series to calculate representative d[NO₂]/dt values, all air parcels must be photochemically equivalent. This requires that (i) air parcels are sampled at the same latitude in order that the time dependence of solar irradiation is the same and (ii) the partitioning between NO and NO₂ depends only on solar flux. The first criterion is met within a few tenths of a degree latitude (Table 1). Note also that the local time for individual air parcels must be used to calculate d[NO₂]/dt, since the variation of NO₂ is a function of sun position. The second criterion requires that the values of O₃, p, and T be nearly constant. All are reasonably constant as indicated in Table 1. In this data set the net contribution of the ClO, BrO, and HO₂ terms in (2) is 8% at 70° SZA decreasing to about 2% at 90° [Del Negro et al., 1999] and, thus generally represents a negligible contribution to the high-SZA analysis here.

NO_x is not sufficiently constant to calculate d[NO₂]/dt directly from the time series data. As an alternative, a normalized derivative is used to calculate d[NO2]/dt. In this case, NO2 is first normalized to NO_x and the quantity $d([NO_2]/[NO_x])/dt$ is calculated. The d[NO₂]/dt term is then calculated by multiplying d([NO₂]/[NO_x])/dt by NO_x. This approach is valid for two reasons. First, NO_x in any given air parcel along the flight track is essentially constant for 85° < SZA < 93° because NO_x photochemical production and loss is small over the associated time interval. Second, since NO₂ equals NO_x in the dark, NO₂/NO_x values just after sunrise or before sunset are equivalent to the fractional change of NO2 from the dark values found just before sunrise or after sunset, respectively. Therefore, NO2/NOx values in different air parcels are independent of absolute NO2 values and inter-comparable. A useful calculation of d[NO₂]/dt using this method requires that a second-order polynomial function be first fitted to the NO₂/NO_x data at high SZA before calculating values of d[NO₂]/dt. The fit prevents random noise in the NO and NO₂ measurements from affecting the calculated d[NO₂]/dt ratio. The total error in $d[NO_2]/dt$ is estimated to be about $\pm 13\%$.

The ratio of calculated d[NO₂]/dt and k₂[O₃][NO] terms is shown in Figure 1. Values of the ratio that exceed 0.1, which typically occur for SZAs greater than about ~90°, indicate the importance of the time-dependent term for photochemically derived J_{NO₂}. Below 90°, the ratio is less than 0.1 and approaches zero as SZA decreases. At the high latitudes of these flights, SZAs change relatively slowly even at high SZAs. At lower latitudes, d[NO₂]/dt generally is more significant compared to k₂[O₃][NO] for 85° < SZA < 90°. The different SZA dependence between sunrise and sunset in Figure 1 is specific to these flights, depending in large part on the differences in p, T, and O₃.

Model calculations. Model JNO2 values were calculated with two independent, spherical, isotropic, multiple-scattering models of the atmospheric radiation field, utilizing total column ozone and surface reflectivity as measured by the Total Ozone Mapping Spectrometer (TOMS). The model results, provided by the Johns Hopkins University Applied Physics Laboratory [Swartz et al., 1999 and references therein] and the Jet Propulsion Laboratory [Salawitch et al., 1994 and references therein], were used in Del Negro et al. [1999] and were referred to as APLTOMS and JPL_{TOMS}, respectively. The uncertainty associated with calculated JNO2 values is estimated to be ±15% based on uncertainties for the radiative flux and NO2 photolysis cross sections. The average difference between the two sets of JNO2 values is about 3% for SZAs less than 85°, and reduces to negligible at 93° SZA. Only APL_{TOMS} results are shown below since JPL_{TOMS} results lead to essentially the same conclusions.

Results and Discussion

Figure 2 illustrates radiative model JNO2 versus photochemically derived JNO2 values, for 89° < SZA < 93°. The photochemical values are shown with the calculated values of d[NO₂]/dt (Figure 2A) and with the steady state approximation (Figure 2B). The inclusion of the d[NO₂]/dt term significantly tightens the correlation between the two sets of JNO2 values. Although this is not a proof of correctness, the tighter correlation in Figure 2A provides a consistency check of the time-dependent method. As shown, the correlation is linear throughout the JNO2 value range. The slope of a linear fit through the origin is about 1.13 (R = 0.99), slightly larger than those reported by Del Negro et al. [1999, plate 4b], but within range of reported data. The average difference between the two sets of JNO2 values (~ 13%) is significantly smaller than the uncertainties in either photochemically derived values (~ 60%, cf. Del Negro et al. [1999]) or radiatively modeled values. The results are also consistent with Bösch et al. [2000], who report good agreement between measured and calculated radiation flux for JNO2 at high SZAs at altitudes between 2.5 and 30 km.

Figure 2A is extended to higher J_{NO_2} values by including the lower SZA data from the four flights (Figure 3). Photochemical J_{NO_2} values derived using the steady state approximation fall in the range of 4 x 10^{-3} s⁻¹ to 0.01 s⁻¹. The correlation with model values is linear over this range of approximately two orders of magnitude. The slope of the fitted line in this case is about 1.11 (R = 0.98), very close to that found with only the higher SZA data. The results presented in Figures 2 and 3 extend the high SZA limit used in *Del Negro et al.* [1999] from 85° to 93° SZA.

For the photochemically derived J_{NO2} values, the precision generally depends on the quality of the measurements used in Equation (2) as well as p and T variations (used in determining reaction rates). Due to the low concentrations of HO₂, ClO, and BrO and slowly varying SZAs in these flights, J_{NO2} in a majority of air parcels can be approximated as

bppmv = parts per million by volume cppbv = parts per billion by volume

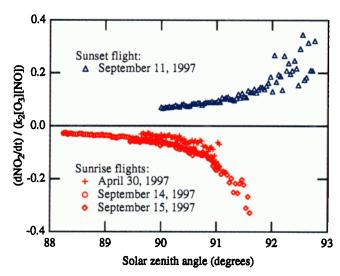


Figure 1. Ratio of d[NO₂]/dt to (k₂[O₃][NO]) as a function of SZA for sunrise (red) and sunset (blue) flights. The analysis is limited at high SZA by the declining signal-to-noise of the NO data.

$$J_{NO_2} \approx k_2[O_3][NO] / [NO_2].$$
 (3)

Because p, T and O_3 for each flight were virtually constant and their measurement uncertainties are small, the precision of a photochemically derived J_{NO_2} value therefore depends primarily on the precision of NO/NO_2 . Furthermore, since the flight-to-flight variations in P and T were small (Table 1), uncertainty in k_2 does not affect the overall precision of photochemically derived J_{NO_2} . For SZAs < 87°, the 1- σ precision of NO/NO_2 is estimated to be 10% and is dominated by the error in NO_2 . For 87° < SZAs < 89° the precision is about 7%, and increases linearly to 20% (NO dominant) at 93° SZA. The estimated 1- σ precision of NO/NO_2 is shown in Figure 3 as dashed lines. As can be seen,

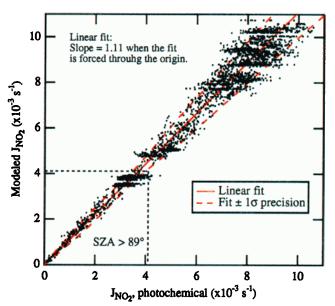


Figure 3. The extension of Figure 2A to higher J_{NO_2} values (70° < SZA < 89°). Data from all four flights are included. The steady-state approximation is used for deriving the points with SZA < 89°. A linear fit gives a slope of 1.11 (R = 0.98) with a forced zero intercept.

they indeed envelop a majority of the photochemically derived J_{NO_2} values.

The model J_{NO_2} values are generally insensitive to albedo and cloud top height for $70^{\circ} < SZAs < 93^{\circ}$ [Swartz et al., 1999]. For the conditions considered here, the changes in modeled J_{NO_2} values are less than 10% and 8% when albedo varies between 0 and 0.6 and terrain/cloud top height from ground to 6 km, respectively. More importantly, the standard deviations of the ratios of J_{NO_2} values calculated with albedo and/or terrain/cloud top height kept at zero to values reported here are less 5% over this

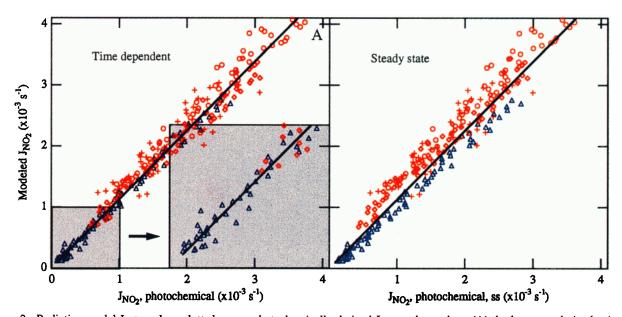


Figure 2. Radiative model J_{NO_2} values plotted versus photochemically derived J_{NO_2} values where (A) the latter are derived using the time-dependent method (Equation (2)) or (B) using the steady state assumption (d[NO₂]/dt = 0). The photochemical data set is the same as that shown in Figure 1. Data for the lowest range of J_{NO_2} are expanded in an inset for (A). Symbols and colors are used the same way as in Figure 1. A linear fit to the data in (A) has a slope of 1.13 with a forced zero intercept. The fit is also shown in (B) as a visual aid.

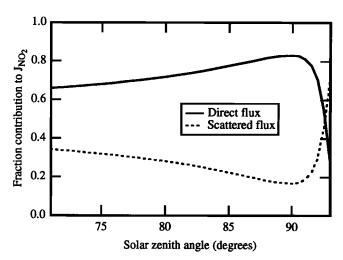


Figure 4. Modeled fractional contributions of direct and scattered flux to JNO₂ as a function of SZA at 20 km.

SZA range. Atmospheric refraction has a negligible effect on J_{NO_2} at SZAs < 90°. Because the wavelength (310 - 410 nm) dependence of J_{NO_2} is also fairly insensitive to SZA for the range considered here and because of constant P and T, the uncertainty in the NO_2 photodissociation cross section does not affect the precision of the model results.

The most significant factor affecting the accuracy of model J_{NO_2} in this SZA region is the amount of scattering of the solar irradiance as a function of SZA. Results of the scattering model shown in Figure 4 show that the contribution to J_{NO_2} of direct solar flux varies from about 65% at 70° to over 80% at 90° and then to less than 30% at 93°. The scattering contribution varies inversely with the direct component. Therefore, the linearity of the correlations in Figures 2 and 3 indicates that the model treatment of the direct and scattered contributions to the radiation field are accurate to approximately 7 - 20% on average over the SZA range of 70° to 93° SZA.

Conclusions

In situ measurements of chemical species and atmospheric parameters were used to derive values of J_{NO_2} in the lower stratosphere using a time-dependent, photochemical method. The method accounts for the departure of NO and NO2 abundances from steady state values. J_{NO_2} values were calculated for the SZA range of 88° to 93° near sunrise and sunset and extended to 70° using the steady state approximation, thereby extending the range of previous studies. The precision of the photochemically derived J_{NO_2} values, shown to be 7 - 20%, is dependent only on the precision of the measured NO/NO2 ratio.

Model J_{NO2} values derived from calculations of direct and scattered solar flux show a linear relationship with photochemically derived values over the full SZA range. The slope of a linear fit is 1.11 (1.13 if only data at SZA > 89° are used). The

agreement of modeled and photochemically derived J_{NO_2} values is consistent with several previous studies. The good linear relationship at high SZA demonstrates that the modeling of scattered light in the 310 - 410 nm wavelength range near sunrise and sunset is accurate to within the precision of measurements used in the photochemical derivation.

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References

Bösch, H., et al., Comparison of measured and modeled stratospheric photolysis frequencies of NO₂ (J_{NO₂}) at large solar zenith angles, Geophys. Res. Lett., in press, 2001.

Del Negro, L. A., et al., Comparison of modeled and observed values of NO₂ and J_{NO₂} during the Photochemistry of Ozone Loss in the Arctic Region in Summer (POLARIS) mission, J. Geophys. Res., 104, 26.687-26.704, 1999.

DeMore, W. B., et al., Chemical kinetics and photochemical data for use in stratospheric modeling, JPL Publ. 97-4, Jet Propul. Lab., Pasadena, Calif., 1997.

Gao, R. S., et al., Partitioning of the reactive nitrogen reservoir in the lower stratosphere of the southern hemisphere: Observations and modeling, J. Geophys. Res., 102, 3935-3949, 1997.

Kawa, S. R., et al., Interpretation of aircraft measurements of NO, ClO, and O₃ in the lower stratosphere, J. Geophys. Res., 95, 18,597-18,609, 1990

Kawa, S. R., et al., Photochemical partitioning of the reactive nitrogen and chlorine reservoirs in the high-latitude stratosphere, J. Geophys. Res., 97, 7905-7923, 1992.

Salawitch, R. J., et al., The diurnal variation of hydrogen, nitrogen, and chlorine radicals: Implications for the heterogeneous production of HNO₂, Geophys. Res. Lett., 21, 2551-2554, 1994.

Swartz, W. H., et al., A sensitivity study of photolysis rate coefficients during POLARIS, J. Geophys. Res., 104, 26,725-26,736, 1999.

Webster, C. R., NO₂/NO partitioning as test of stratospheric ClO concentration over Antarctica, *Geophys. Res. Lett.*, 14, 704-706, 1987.

Wennberg, P. O., et al., Twilight observations suggest unknown sources of OH_x, Geophys. Res. Lett., 26, 1373-1376, 1999.

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